

## خصوصیات کانی شناسی و ژئوشیمیایی توده نفوذی خضرآباد ( شمال غرب تفت )

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**چکیده:** توده نفوذی خضرآباد در شمال غرب تفت رخنمون دارد. به نظر می‌رسد این توده نفوذی از سنگهای دگرگونی اطراف و به ویژه از سنگهای آهکی در برگیرنده به سن کرتاسه زیرین جوانتر باشد و احتمالاً دارای سن الیگو-میوسن است. فراوانترین سنگهای توده نفوذی عبارتند از گرانوودیوریت، گرانیت، کوارتزمونز و دیوریت، کوارتزدیوریت و به مقدار کمتر تونالیت، کوارتزسیینیت و سیینیت. ضمناً کلیه سنگها غنی‌شدگی از عناصر Rb, K, Ba و تهی‌شدگی ازNb, Sr, Ti را نشان می‌دهند.

از نظر زمین شناسی اقتصادی کانی‌سازی مرمر، اسکارن، آهن - مس - سرب - روی و کانی‌های غیرفلزی نظیر کائولینیت قابل ملاحظه است. ژئوترموبارومتری کانی‌های تشکیل دهنده این توده گرانیتوئیدی، دمای ۸۱۰ تا ۹۸۵ درجه سانتی‌گراد و فشار ۲/۴۳ تا ۶/۲ کیلوبار را نشان می‌دهد.

**واژه‌های کلیدی:** گرانیتوئید، گرانیت، دیوریت، کانی‌سازی، ژئوترموبارومتری.



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## Mineralogical and Geochemical characteristics of "Khezr-Abad pluton" NW of Taft, Iran.

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**Abstract:** Khezr-Abad pluton is cropped out in North-West of Taft. This granitoid seems to be younger than the surrounding metamorphosed rocks, particularly the lower cretaceous limestones and probably is implaced during Oligo-Miocene. The most volumetric abundances of the igneous rocks are: granodiorite, granite, quartzmonzo-diorite, quartz-diorite, and in a lesser amount tonalite, quartz-syenite and syenite. All granitoid rocks show Ba, K, Rb enrichment, and Nb, Sr, Ti depletion.

From the economical potential point of view, mineralization of marble, Fe-Cu-Pb-Zn skarn and non-metalic minerals such as kaolinite are considerable.

Geothermobarometry of rock forming minerals of this pluton indicates temperature of 810- 985 °C and pressures of 2.43 – 6.2 kilobars (kbar).

**Keywords:** *Granitoid, Granite, Diorite, Mineralization, Geothermobarometry.*

## 1. Introduction

Khezr-Abad pluton is situated in Nw of Taft (Central Iran), With a  $53^{\circ}30'$  to  $54^{\circ}20'$  eastern longitude and  $31^{\circ}30'$  to  $31^{\circ}52'$  northern latitude (Fig. 1).

With respect to existance of younger plutonic rocks (eg. Khezr-Abad) than shir-kuh pluton in central Iran, the aim of this study is to indicate: relative ageing, chemical compositions and geothermobarometry of related rocks and minerals.

Granitoid rocks were intruded into volcanic rocks (Eocene) and cretaceous limestones (Taft Formation). With respect to the metamorphosed cretaceous limestones and skarn formation, the age of the plutonism of the area are younger than Cretaceous and Eocene (Oligo-Miocene). On the basis of structural sedimentary units division of Iran [1], the pluton is located within the central-Iran zone.

Using discriminant diagrams of Maniar&Piccoli [2], granitoids could be generally divided into three groups: 1-CCG<sup>1</sup>, CAG<sup>2</sup> and IAG<sup>3</sup> 2-RRG<sup>4</sup> and CEUG<sup>5</sup>, and 3-POG<sup>6</sup>.

With this respect the granitoids are post-orogenic and syn-orogenic. From the economical point of view, mineralization of marbles, Fe-Cu-Pb-Zn skarn and non-metalic minerals such as kaolinite are mentionable. Pegmatites in relation to the plutonic rocks are rare with lack of economical value.

## 2. Method of Study

In order to undertake, mineralogical and geochemical studies and determination of pressure and temperature (P&T) conditions of pluton, dense sampling and field works considerations were made and 17 samples of plutonic rocks were selected for major and trace elements analyses by XRF. In addition, some component minerals of plutonic rocks studied by Electron Probe Micro Analysis method (EPMA)\* [at New Brownswick university, Canada, [1998]], and on this basis, P-T conditions of crystallization was estimated. P-T conditions of plutonic emplacement also were calculated, using different geothermobarometers.

<sup>1</sup> Continental collision Granitoids

<sup>2</sup> Continental Arc Granitoids

<sup>3</sup> Island Arc Granitoids

<sup>4</sup> Rift - related Granitoids

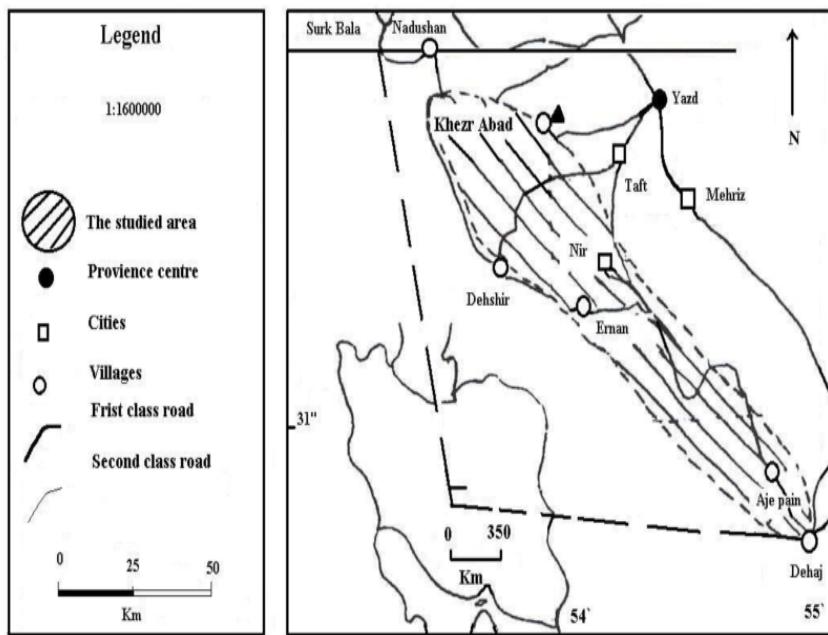
<sup>5</sup> Continental Epriogenic Uplift Granitoids

<sup>6</sup> Post Orogenic Granitoids

\* Name and conditions of instrument:

JEOL superprobe 733 Wavelength Dispersive Spectroscopy (WDS).

Beam current, 10 mA, 15 Kev, Count time, 60 seconds.



**Fig. 1** Geographical position of granitoid of Khezr-Abad (NW, Taft, central Iran).  
 ▲: Khezr-Abad

### 3. Petrography and Mineralogy

Khezr-Abad pluton consist of different ellipsoidal and spherical dark xenoliths, acidic and doleritic daykes, veins of aplite and pegmatite were seen within the granites. Small volcanic masses such as dacite, rhyodacite doms, trachytic and andesitic tuffs and lavas and rarely basaltic-andesite in contact with granites and close to the faults (particularly Dehshir–Baft fault with a trend of NW-SE) are reported. The granitoid rocks under study in area are, granodiorite, granite, quartz monzo-diorite, quartz-diorite, and in a lesser amount tonalite, quartz-syenite and syenite.

Texture of granitoid rocks are granular, granophyric (graphic), perthite, anti-perthite and sieve textures. The graphic textures indicate eutectic or cotectic crystallization of quartz and alkali-feldspars. Table 1 shows chemical analysis of plutonic rocks.

**Table 1** Chemical analysis of plutonic rocks of Khezr-Abad by XRF method.  
Oxides (wt%), traces (ppm).

|       | AD27  | AD26  | AD15  | AD14  | AD12  | AD10  | AD6   | gkcl7 | gk25  | gk22  | gk18  | gk14  | gk12  | gk8   | gk1                            | Oxides/<br>No. Sample |
|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|--------------------------------|-----------------------|
| 62.1  | 59.7  | 60.38 | 62.12 | 73.29 | 74.06 | 56.10 | 69.15 | 62.39 | 74.77 | 68.63 | 58.24 | 66.54 | 67.65 | 69.10 | SiO <sub>2</sub>               |                       |
| 14.57 | 13.98 | 15.82 | 15.94 | 13.28 | 13.57 | 13.73 | 13.87 | 15.25 | 13.08 | 13.96 | 15.13 | 13.97 | 13.89 | 13.89 | Al <sub>2</sub> O <sub>3</sub> |                       |
| 5.65  | 7.18  | 6.5   | 5.31  | 0.86  | 0.84  | 7.38  | 2.86  | 3.37  | 0.26  | 3.53  | 6.37  | 3.55  | 3.38  | 2.85  | CaO                            |                       |
| 2.91  | 2.11  | 2.28  | 2.90  | 2.96  | 6.13  | 2.26  | 3.88  | 8.21  | 3.86  | 3.69  | 1.94  | 3.68  | 3.51  | 3.84  | K <sub>2</sub> O               |                       |
| 2.24  | 2.37  | 2.36  | 2.04  | 1.69  | 0.98  | 2.49  | 1.89  | 1.99  | 0.86  | 1.99  | 2.41  | 2.00  | 1.9   | 1.88  | Fe <sub>2</sub> O <sub>3</sub> |                       |
| 3.86  | 4.81  | 4.2   | 2.74  | 0.74  | -     | 6.13  | 1.26  | 1.18  | -     | 1.53  | 5.52  | 1.52  | 1.2   | 1.26  | FeO                            |                       |
| 2.93  | 2.71  | 3.49  | 3.16  | 4.44  | 3.20  | 2.67  | 3.60  | 2.96  | 4.88  | 3.46  | 3.98  | 3.46  | 3.47  | 3.61  | Na <sub>2</sub> O              |                       |
| 2.92  | 3.04  | 3.31  | 2.04  | 0.57  | 0.22  | 4.16  | 1.29  | 1.39  | 0.28  | 1.42  | 3.65  | 1.42  | 1.38  | 1.3   | MgO                            |                       |
| 0.7   | 0.81  | 0.81  | 0.54  | 0.19  | 0.08  | 0.99  | 0.39  | 0.49  | 0.18  | 0.44  | 0.86  | 0.45  | 0.4   | 0.38  | TiO <sub>2</sub>               |                       |
| 0.185 | 0.21  | 0.26  | 0.16  | 0.064 | 0.027 | 0.265 | 0.114 | 0.173 | 0.114 | 0.143 | 0.26  | 0.141 | 0.125 | 0.11  | P <sub>2</sub> O <sub>5</sub>  |                       |
| 0.112 | 0.124 | 0.128 | 0.084 | 0.024 | 0.013 | 0.169 | 0.063 | 0.088 | 0.008 | 0.086 | 0.19  | 0.07  | 0.061 | 0.064 | MnO                            |                       |
| 98.08 | 97.04 | 99.54 | 97.04 | 98.10 | 99.12 | 96.34 | 98.37 | 97.49 | 98.29 | 98.86 | 98.55 | 98.8  | 96.96 | 98.31 | Total                          |                       |

Table 1(cont.)

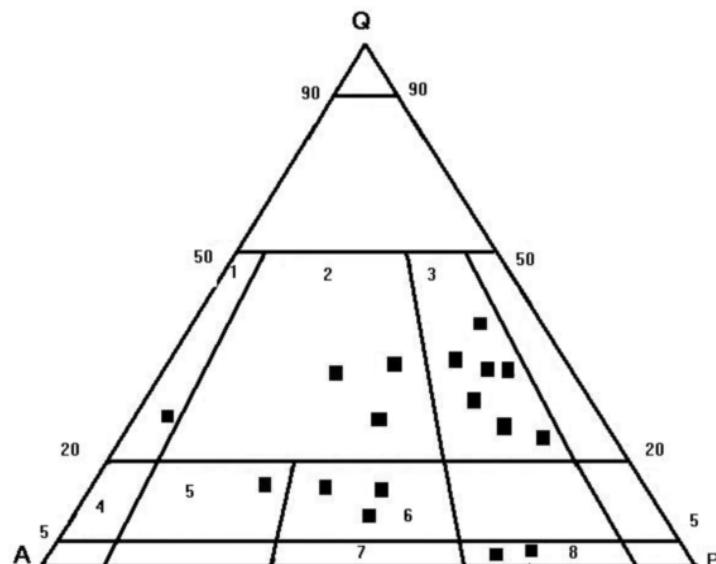
| AD39  | AD37  |
|-------|-------|
| 57.65 | 58.47 |
| 14.46 | 13.76 |
| 6.12  | 10.12 |
| 2.73  | 6.87  |
| 2.54  | 2.21  |
| 5.72  | 3.29  |
| 2.92  | 1.18  |
| 3.86  | 1.78  |
| 1.01  | 0.67  |
| 0.25  | 0.159 |
| 0.16  | 0.186 |
| 97.42 | 98.66 |

Table 1 (cont.)

| AD14 | AD12 | AD10 | AD6 | gkcl7 | gk25 | gk22 | gk18 | gk14 | gk12 | gk8 | gk1 | Trace.E/<br>No.sample |
|------|------|------|-----|-------|------|------|------|------|------|-----|-----|-----------------------|
| 152  | 65   | 56   | 141 | 136   | 157  | 105  | 145  | 175  | 144  | 126 | 137 | Zr                    |
| 290  | 106  | 48   | 296 | 173   | 280  | 25   | 195  | 200  | 199  | 183 | 175 | Sr                    |
| 103  | 130  | 188  | 94  | 150   | 193  | 110  | 131  | 125  | 134  | 138 | 146 | Rb                    |
| 11   | 7    | 21   | 20  | 16    | 16   | 9    | 18   | 15   | 18   | 14  | 18  | Pb                    |
| 58   | 23   | 13   | 108 | 31    | 30   | 10   | 42   | 73   | 41   | 36  | 32  | Zn                    |
| 33   | 6    | 4    | 40  | 7     | 15   | 1    | 9    | 19   | 8    | 0   | 8   | Cu                    |
| 5    | 3    | 4    | 9   | 10    | 4    | 0    | 11   | 10   | 10   | 0   | 10  | Ni                    |
| 40   | 84   | 28   | 49  | 46    | 11   | 14   | 44   | 83   | 44   | 50  | 45  | Ce                    |
| 34   | 27   | 20   | 28  | 29    | 8    | 28   | 8    | 43   | 9    | 17  | 31  | Nd                    |
| 51   | 48   | 39   | 56  | 48    | 42   | 32   | 38   | 71   | 37   | 46  | 48  | Nb                    |
| 965  | 144  | 142  | 618 | 390   | 467  | 661  | 675  | 761  | 671  | 351 | 388 | Cl                    |
| 19   | 44   | 57   | 31  | 52    | 19   | 22   | 37   | 25   | 26   | 36  | 50  | La                    |
| 8    | 24   | 41   | 7   | 16    | 5    | 22   | 14   | 7    | 13   | 11  | 17  | Th                    |
| 114  | 183  | 0    | 390 | 119   | 131  | 460  | 395  | 650  | 405  | 44  | 114 | F                     |
| 91   | 29   | 16   | 175 | 56    | 58   | 12   | 59   | 123  | 58   | 59  | 55  | V                     |
| 70   | 6    | 81   | 34  | 102   | 8    | 0    | 66   | 41   | 68   | 62  | 103 | Cr                    |
| 431  | 155  | 60   | 341 | 503   | 960  | 269  | 523  | 299  | 525  | 460 | 499 | Ba                    |

|     | AD39 | AD37 | AD27 | AD26 | AD15 |
|-----|------|------|------|------|------|
| 143 | 175  | 150  | 142  | 167  |      |
| 315 | 255  | 278  | 321  | 328  |      |
| 116 | 177  | 93   | 66   | 87   |      |
| 14  | 100  | 19   | 22   | 23   |      |
| 84  | 96   | 71   | 73   | 81   |      |
| 9   | 27   | 26   | 37   | 12   |      |
| 7   | 3    | 9    | 9    | 12   |      |
| 37  | 84   | 61   | 42   | 57   |      |
| 27  | 33   | 17   | 21   | 14   |      |
| 60  | 49   | 45   | 54   | 55   |      |
| 560 | 210  | 755  | 864  | 515  |      |
| 33  | 15   | 18   | 13   | 36   |      |
| 11  | 11   | 9    | 7    | 10   |      |
| 321 | 685  | 737  | 771  | 350  |      |
| 141 | 112  | 132  | 160  | 122  |      |
| 17  | 18   | 67   | 21   | 59   |      |
| 352 | 632  | 408  | 406  | 497  |      |

On the basis of modal classification [3] and Q-P diagram [4], rocks of the area are mainly granodiorite (gk14, gk18, AD10, AD12, AD14, AD15, AD26, AD27), granite (gk1, gk22, gkc17), quartz-monzonodiorite (gk12), quartz-diorite (gk25, AD22), and to a lesser amount tonalite (gk 8), syenite (AD37) and quartz-syenite (AD39) (Figs. 2 and 3).



**Fig 2** Streckeisen diagram (1982) Showing plot of igneous rocks of the studied area.

1- Alkali feldspar granite

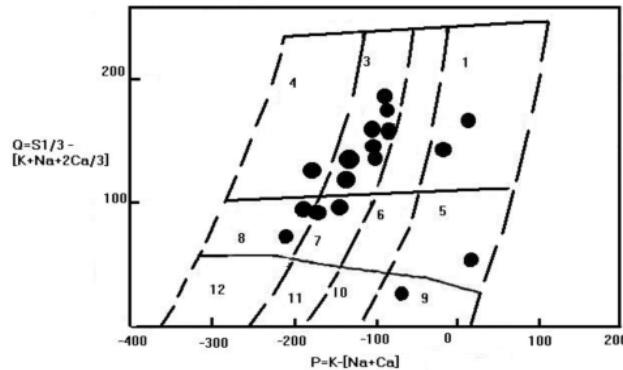
2- Granite

3- Granodiorite

5- Quartz-syenite

6- Quartz-monzonite

8- Monzo- diorite



**Fig. 3** Q-P diagram after Debon & Lefort [4], Showing plot of igneous rocks of the studied area. 1- Granite, 3- Granodiorite, 4- Tonalite, 5- Quartz-syenite, 7- Quartz-monzo diorite, 8- Quartz-diorite, 9- Syenite.

Minerals of intrusive rocks are quartz, plagioclase (albite, oligoclase and andesine), orthoclase, pyroxene, amphibole and mica (biotite and to a lesser amount muscovite). Minor mineral constituents of these rocks are apatite, zircon (as inclusions in biotite), sphene, tourmaline, spinel (hercynite), hematite and magnetite. The above minerals are firstly distinguished under the microscope, then by XRD method and finally by EPMA method which are described as the following.

### 3.1. Pyroxenes

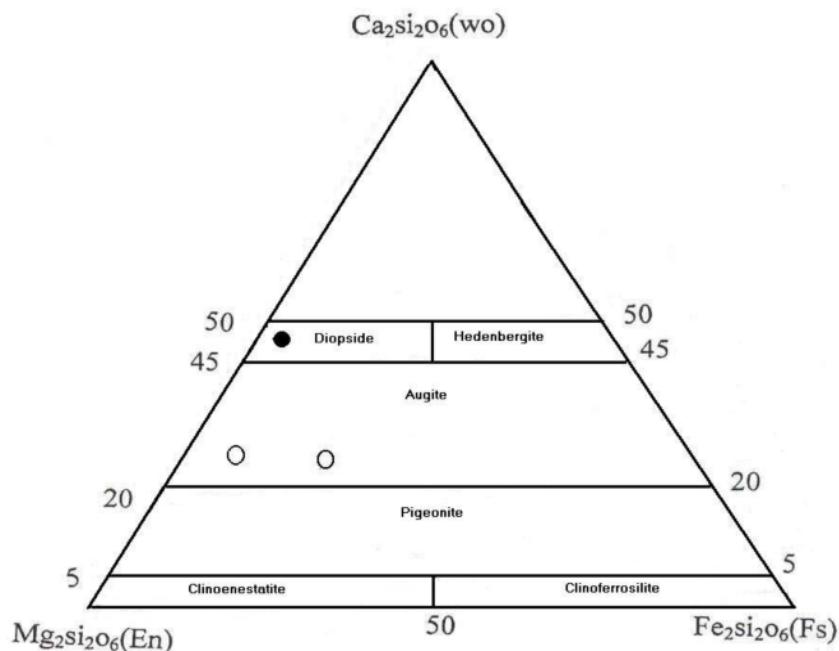
Pyroxenes in plutonic rocks are of diopside and augite types. Aegirine-augite pyroxene in some more sodic magmatic rocks are found. According to minerals nomenclature presented by Morimoto [5], the pyroxens are of calcic type (diopside, augite), (Fig. 4). Table 2 shows chemical analysis of pyroxenes. Photomicrograph No. 1 shows a microscopic view of minerals in a granitic rock of the area.

**Table 2** Representatives of pyroxene minerals.

| GK14  | AD34  | GK17  | Rock                           |
|-------|-------|-------|--------------------------------|
| CPX   | CPX   | CPX   | Mineral                        |
| 50.45 | 53.46 | 56.45 | SiO <sub>2</sub>               |
| 1.56  | 0.12  | 0.07  | TiO <sub>2</sub>               |
| 6.98  | 0.09  | 2.5   | Al <sub>2</sub> O <sub>3</sub> |
| 16.09 | 19.21 | 22.6  | MgO                            |
| 11.41 | 25.58 | 13.38 | CaO                            |
| 0.14  | 0     | 0.08  | Cr <sub>2</sub> O <sub>3</sub> |
| 0.43  | 0.18  | 0.06  | MnO                            |
| 11.2  | 1.28  | 3.77  | FeO <sub>1</sub>               |
| 1.64  | 0     | 0.7   | Na <sub>2</sub> O              |
| 0     | 0     | 0     | K <sub>2</sub> O               |
| -     | -     | -     | F                              |
| -     | -     | -     | Cl                             |
| 99.9  | 99.92 | 99.61 | Total                          |



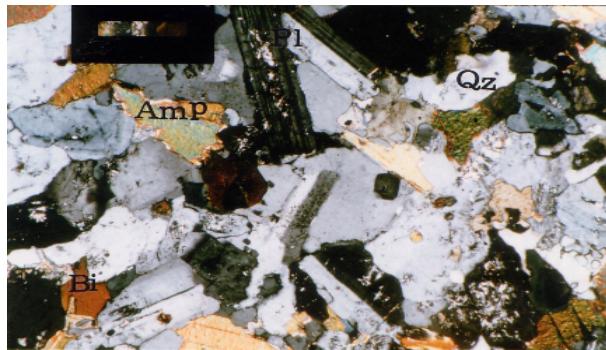
**Photomicrograph 1** showing amphibole, biotite and plagioclase crystals in granitoid rocks. Length of marker is 5 mm. Amphibole= Amp, Biotite= Bi, Plagioclase= PL.



**Fig. 4** Field of Ca, Mg and Fe CPX, according to ref. [5].

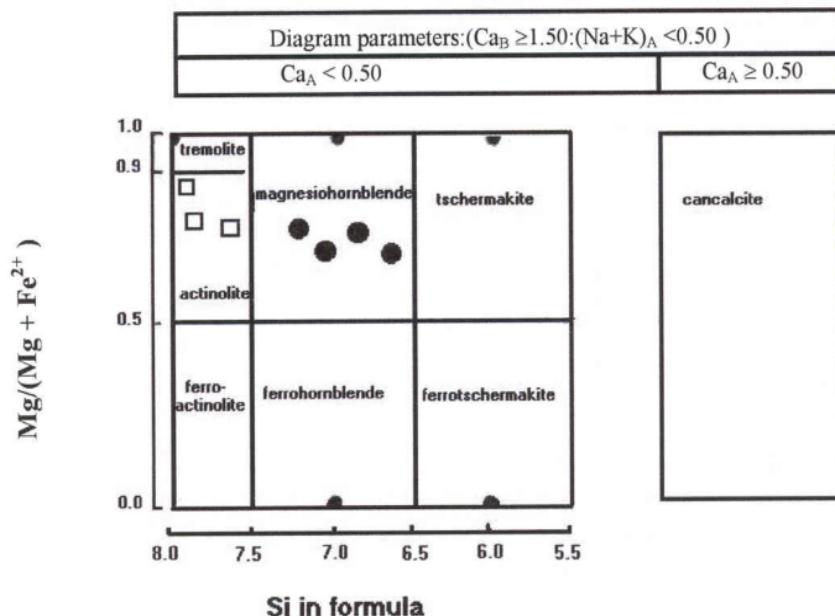
### 3.2. Amphiboles

Amphiboles in granitoid rocks coexist with pyroxene (augite), plagioclase, mica (biotite), orthoclase and oxides. Amphiboles in these rocks are actinolite (in granodiorite and granite), edenite (in andesite), and magnesiohornblende (in granodiorite and quartz-diorite) types. Photomicrograph No. 2 shows a view of a thin section of a plutonic rock.



**Photomicrograph 2** showing crystals of amphibole, biotite, plagioclase and quartz within the granitoid rocks. Length of marker is 5 mm. Amphibole= Amp, Biotite= Bi, Plagioclase= Pl, Quartz= Qz.

According to Leake [6], amphiboles within the granitoid rocks are mainly of calcic type, actinolite, edenite, magnesiohornblende which is shown in Fig. 5. Table 3 shows EPMA results of some amphiboles from granitoid rocks of Khezr-Abad.



**Fig. 5** Classification of calcic amphiboles (after Leake, [6]). Amphiboles are of actinolite and magnesiohornblende types.

- In granodiorite and quartz-diorite (AD14, AD22).
- In granodiorite and granite (gk18, gk1).

**Table 3** Chemical analysis of amphiboles by EPMA method. Cations calculation is on the basis of 23 (O, F, Cl).

| Amphibole<br>No                | Actinolite      |           |         | Actinolite       |         |           | Magnesiohornblende |           |         | Magnesiohornblende |         |       |
|--------------------------------|-----------------|-----------|---------|------------------|---------|-----------|--------------------|-----------|---------|--------------------|---------|-------|
|                                | gk <sub>1</sub> |           |         | gk <sub>18</sub> |         |           | AD <sub>14</sub>   |           |         | AD <sub>22</sub>   |         |       |
|                                | Oxides          | (wt)<br>% | Cations | (wt)<br>%        | Cations | (wt)<br>% | Cations            | (wt)<br>% | Cations | (wt)<br>%          | Cations |       |
| SiO <sub>2</sub>               | 54.01           | Si        | 7.921   | 51.9             | Si      | 7.757     | 47.13              | Si        | 7.3     | 48.34              | Si      | 7.29  |
| TiO <sub>2</sub>               | 0.29            | A14       | 0.079   | 0.53             | A14     | 0.243     | 0.38               | A14       | 0.69    | 1.02               | A14     | 0.706 |
| Al <sub>2</sub> O <sub>3</sub> | 8.3             | A16       | 1.349   | 7.48             | A16     | 1.069     | 7.54               | Al'       | 0.66    | 8.81               | A16     | 0.85  |
| FeO <sub>t</sub>               | 11.41           | Ti        | 0.032   | 13.52            | Ti      | 0.059     | 16.18              | Ti        | 0.43    | 15.32              | Ti      | 0.115 |
| MnO                            | 1.19            | Fe        | 1.393   | 0.92             | Fe      | 1.689     | 0.91               | Fe        | 2.09    | 0.89               | Fe      | 1.93  |
| MgO                            | 10.66           | Mg        | 2.328   | 11.07            | Mg      | 2.463     | 11.91              | Mg        | 2.74    | 11.85              | Mg      | 2.65  |
| CaO                            | 10.68           | Mn        | 0.147   | 11.29            | Mn      | 0.107     | 11.25              | Mn        | 0.12    | 11.3               | Mn      | 0.11  |
| Cr <sub>2</sub> O <sub>3</sub> | 0.01            | Ca        | 1.675   | 0.01             | Ca      | 1.806     | 0.1                | Ca        | 1.86    | 0.01               | Ca      | 1.82  |
| Na <sub>2</sub> O              | 1.5             | Cr        | 0.001   | 0.81             | Cr      | 0.001     | 1.39               | Cr        | 0.012   | 1.41               | Cr      | 0.011 |
| K <sub>2</sub> O               | 0.38            | Na        | 0.423   | 0.49             | Na      | 0.233     | 0.44               | Na        | 0.409   | 0.19               | Na      | 0.411 |
| F                              | 0.28            | K         | 0.07    | 0.31             | K       | 0.089     | MI                 | K         | 0.085   | 0.12               | K       | 0.036 |
| Cl                             | 0.09            | F         | 0.123   | 0.09             | F       | 0.143     | 0.09               | F         | 0.149   | 0.05               | F       | 0.057 |
| <b>Total</b>                   | 98.8            | C1        | 0.022   | 98.42            | C1      | 0.022     | 97.63              | C1        | 0.023   | 99.31              | C1      | 0.012 |

### 3.3. Micas

Micas in the magmatic rocks of the area are brown biotites and within the metamorphic rocks (skarns) are phlogopite. These are confirmed by EPMA analysis. Biotites and phlogopites show a wide range of variations in Al and Mg contents, Al<sub>2</sub>O<sub>3</sub> (wt %) variations are from 15.38% in biotite to 18.16% in phlogopite, MgO also varies from 6.14% in biotite to 25.25% in phlogopite. The results of four selected samples from biotite are presented in Table 4.

**Table 4** EPMA results of biotite from the granitoid rocks. Cations calculation is based on 23 (O, F, Cl ).

| Mica<br>No                     | Biotite         |                 |         | Biotite          |                 |           | Biotite          |                 |         | Biotite          |                 |        |
|--------------------------------|-----------------|-----------------|---------|------------------|-----------------|-----------|------------------|-----------------|---------|------------------|-----------------|--------|
|                                | gk <sub>1</sub> |                 |         | gk <sub>18</sub> |                 |           | AD <sub>14</sub> |                 |         | AD <sub>22</sub> |                 |        |
|                                | Oxides          | (wt)<br>%       | Cations | (wt)<br>%        | Cations         | (wt)<br>% | Cations          | (wt)<br>%       | Cations | (wt)<br>%        | Cations         |        |
| SiO <sub>2</sub>               | 35.41           | Si              | 5.855   | 35.08            | Si              | 5.877     | 35.47            | Si              | 5.861   | 36.4             | Si              | 5.99   |
| TiO <sub>2</sub>               | 2.73            | A1 <sup>4</sup> | 2.145   | 1.82             | A1 <sup>4</sup> | 2.123     | 2.72             | A1 <sup>4</sup> | 2.139   | 3.23             | A1 <sup>4</sup> | 2.01   |
| Al <sub>2</sub> O <sub>3</sub> | 17.41           | Al <sup>6</sup> | 1.235   | 17.13            | Al <sup>6</sup> | 1.258     | 17.27            | Al <sup>6</sup> | 1.218   | 15.38            | Al <sup>6</sup> | 0.96   |
| FeO <sub>t</sub>               | 20.57           | Ti              | 0.338   | 24.05            | Ti              | 0.231     | 22.21            | Ti              | 0.337   | 19.87            | Ti              | 0.396  |
| MnO                            | 0               | Fe              | 2.843   | 0.72             | Fe              | 3.371     | 0                | Fe              | 3.069   | 0                | Fe              | 2.732  |
| MgO                            | 7.94            | Mn              | 0       | 6.14             | Mn              | 0.101     | 7.02             | Mn              | 0       | 9.65             | Mn              | 0      |
| CaO                            | 0               | Mg              | 1.948   | 0                | Mg              | 1.530     | 0                | Mg              | 1.728   | 0                | Mg              | 2.366  |
| Na <sub>2</sub> O              | 0               | Ca              | 0       | 0.3              | Ca              | 0         | 0                | Ca              | 0       | 0                | Ca              | 0      |
| K <sub>2</sub> O               | 9.94            | Na              | 0       | 9.81             | Na              | 0.101     | 9.86             | Na              | 0       | 9.74             | Na              | 0      |
| F                              | 1.28            | K               | 2.087   | 1.92             | K               | 2.093     | 1.35             | K               | 2.066   | 1.21             | K               | 2.039  |
| C1                             | 0.09            | F               | 0.666   | 0.74             | F               | 1.016     | 0.02             | F               | 0.705   | 0.05             | F               | 0.623  |
| BaO                            | 0.26            | C1              | 0.0248  | 0                | C1              | 0.211     | 0.38             | C1              | 0.0055  | 0.25             | C1              | 0.0138 |
| <b>Total</b>                   | 95.63           | Ba              | 0.0159  | 97.71            | Ba              | 0         | 96.3             | Ba              | 0.023   | 95.78            | Ba              | 0.0158 |

### 3.4. Plagioclases

Magmatic rocks of the area, both volcanic and plutonic include plagioclases. Quartz is the first and plagioclase is the second in the order of abundances in both volcanic and plutonic rocks of the area. Seventeen plagioclases are analysed by EPMA. The result is shown in Table 5. Plagioclases are oligoclase to andesine in the acidic and intermediate rocks and labradorite in the basic rocks.

Plagioclases have been analysed from core to rim. Some of them showing oscillatory and reverse zoning, which may be as a result of variations in physical conditions (variations in partial pressure of water vapour ( $H_2O$ ) and temperature [7]. In some cases anorthite content of plagioclase decrease from core to rim, but suddenly increase in rim. This may be occur as a result of variations in chemical composition of magma. Increasing of Ca content of magma may be occur as a result of assimilation of limestone by magma during rising to the surface.

**Table 5** EPMA results of plagioclases from granitoid rocks. Cations calculation is based on 32 Oxygens.

| Plagioclase                    |                        | Oligoclase             |                  | Andesine                  |                          | Andesine                |                         | Oligoclase              |                         | Oligoclase              |          | Andesine |          | Andesine |  |
|--------------------------------|------------------------|------------------------|------------------|---------------------------|--------------------------|-------------------------|-------------------------|-------------------------|-------------------------|-------------------------|----------|----------|----------|----------|--|
| No                             | gk <sub>1</sub><br>(c) | gk <sub>1</sub><br>(r) | gk <sub>18</sub> | gk <sub>18</sub><br>(c 1) | gk <sub>18</sub><br>(C2) | gk <sub>18</sub><br>(r) | AD <sub>14</sub><br>(c) | AD <sub>14</sub><br>(r) | AD <sub>22</sub><br>(c) | AD <sub>22</sub><br>(r) | Andesine | Andesine | Andesine | Andesine |  |
| Oxides                         |                        |                        |                  |                           |                          |                         |                         |                         |                         |                         |          |          |          |          |  |
| SiO <sub>2</sub>               | 60.95                  | 62.23                  | 60.06            | 56.3                      | 1                        | 60.8                    | 60.38                   | 61.64                   | 61.25                   | 58.87                   | 60.24    |          |          |          |  |
| TiO <sub>2</sub>               | 0                      | 0                      | 0                | 0                         | 0                        | 0                       | 0                       | 0                       | 0                       | 0                       | 0        |          |          |          |  |
| Al <sub>2</sub> O <sub>3</sub> | 24.69                  | 24                     | 25.73            | 27.12                     | 24.31                    | 25.43                   | 24.76                   | 24.79                   | 27.55                   | 25.55                   |          |          |          |          |  |
| FeO <sub>t</sub>               | 0.08                   | 0.09                   | 0.17             | 0.15                      | 0                        | 0.09                    | 0.06                    | 0.08                    | 0.21                    | 0.19                    |          |          |          |          |  |
| MgO                            | 0                      | 0                      | 0                | 0.06                      | 0                        | 0                       | 0                       | 0                       | 0                       | 0                       |          |          |          |          |  |
| CaO                            | 5.12                   | 4.53                   | 6.29             | 8.01                      | 4.79                     | 5.91                    | 5.19                    | 5.09                    | 8.95                    | 6.41                    |          |          |          |          |  |
| Na <sub>2</sub> O              | 8.36                   | 8.75                   | 7.67             | 7.48                      | 8.91                     | 7.93                    | 8.63                    | 8.52                    | 6.14                    | 7.82                    |          |          |          |          |  |
| K <sub>2</sub> O               | 0.35                   | 0.47                   | 0                | 0.17                      | 0.32                     | 0.38                    | 0.17                    | 0.36                    | 0.15                    | 0.11                    |          |          |          |          |  |
| <b>Total</b>                   | 99.55                  | 100.1                  | 99.92            | 99.32                     | 99.13                    | 100.1                   | 100.5                   | 100.1                   | 101.9                   | 100.3                   |          |          |          |          |  |
| <i>cations</i>                 |                        |                        |                  |                           |                          |                         |                         |                         |                         |                         |          |          |          |          |  |
| Si                             | 10.87                  | 11.036                 | 10.844           | 10.204                    | 10.902                   | 10.737                  | 10.887                  | 10.873                  | 10.337                  | 10.696                  |          |          |          |          |  |
| Al                             | 5.188                  | 5.006                  | 5.476            | 5.793                     | 5.133                    | 5.325                   | 5.162                   | 5.185                   | 5.702                   | 5.337                   |          |          |          |          |  |
| Fe                             | 0.01                   | 0.01                   | 0.022            | 0.0217                    | 0                        | 0.0128                  | 0.0088                  | 0.011                   | 0.03                    | 0.027                   |          |          |          |          |  |
| Mg                             | 0                      | 0                      | 0                | 0.015                     | 0                        | 0                       | 0                       | 0                       | 0                       | 0                       |          |          |          |          |  |
| Na                             | 2.894                  | 3.003                  | 1.768            | 2.61                      | 3.084                    | 2.716                   | 2.952                   | 2.923                   | 2.09                    | 2.69                    |          |          |          |          |  |
| Ca                             | 0.975                  | 0.852                  | 1.215            | 1.546                     | 0.916                    | 1.122                   | 0.977                   | 0.96                    | 1.678                   | 1.216                   |          |          |          |          |  |
| K                              | 0.0857                 | 0.106                  | 0                | 0.0217                    | 0.071                    | 0.085                   | 0.038                   | 0.081                   | 0.033                   | 0.023                   |          |          |          |          |  |
| Mol percent                    | <b>Ab</b>              | 7118                   | 75.82            | 59                        | 62.48                    | 75.75                   | 69.23                   | 74.41                   | 73.74                   | 54.98                   | 68.46    |          |          |          |  |
|                                | <b>An</b>              | 24.65                  | 21.52            | 41                        | 37                       | 22.5                    | 28.60                   | 24.63                   | 24.22                   | 44.15                   | 30.95    |          |          |          |  |
|                                | <b>Or</b>              | 2.17                   | 2.67             | 0                         | 0.52                     | 1.75                    | 2.17                    | 0.96                    | 2.04                    | 0.87                    | 0.59     |          |          |          |  |

### 3.5. Feldspars

Basically feldspars are of a series of solid solutions between  $KAlSi_3O_8$  and  $NaAlSi_3O_8$  with a little amount of  $CaAl_2Si_2O_8$ . In general anorthite content is less than 5% for a composition between  $Or_{100}Ab_0$  to  $Or_{60}Ab_{40}$ , but in Na end-members it is a little higher. Alkali feldspars also like plagioclase is a main constituent of granitoid rocks. Alkali feldspars are of orthoclase, microcline and albite type. Six samples of feldspars were analysed by EPMA, which are presented in Table 6.

**Table 6** EPMA results of alkali feldspars from plutonic rocks . Cations calculation is based on 32 Oxygens.

| Alkali feldspar                    | orthoclase      | microcline              | Albite                  | orthoclase              | orthoclase       | orthoclase       |
|------------------------------------|-----------------|-------------------------|-------------------------|-------------------------|------------------|------------------|
| No                                 | gk <sub>1</sub> | gk <sub>18</sub><br>(1) | gk <sub>18</sub><br>(2) | Gk <sub>18</sub><br>(3) | AD <sub>14</sub> | AD <sub>22</sub> |
| <b>Oxides</b>                      |                 |                         |                         |                         |                  |                  |
| <b>SiO<sub>2</sub></b>             | 64.02           | 65.32                   | 67.3                    | 63.9                    | 63.1             | 63.25            |
| <b>TiO<sub>2</sub></b>             | 0.07            | 0                       | 0                       | 0.09                    | 0                | 0                |
| <b>Al<sub>2</sub>O<sub>3</sub></b> | 19.22           | 19.11                   | 21.11                   | 18.87                   | 18.14            | 18.27            |
| <b>FeO<sub>t</sub></b>             | 0.05            | 0                       | 0                       | 0.08                    | 0.04             | 0.12             |
| <b>MgO</b>                         | 0               | 0                       | 0                       | 0.07                    | 0.06             | 0.06             |
| <b>CaO</b>                         | 0.66            | 0                       | 0.49                    | 0.6                     | 0.52             | 0.83             |
| <b>Na<sub>2</sub>O</b>             | 3.11            | 1.59                    | 10.01                   | 3.4                     | 2.55             | 3.68             |
| <b>K<sub>2</sub>O</b>              | 12.61           | 14.32                   | 1.26                    | 12.75                   | 12.81            | 12.94            |
| <b>BaO</b>                         | 0               | 0                       | 0                       | 0.28                    | 0.53             | 0.07             |
| <b>Total</b>                       | 100.19          | 1                       | 100.17                  | 100.04                  | 97.85            | 99.3             |

#### Cations

|                     |           |        |        |        |        |        |
|---------------------|-----------|--------|--------|--------|--------|--------|
| <b>Si</b>           | 11.779    | 11.937 | 11.774 | 11.787 | 11.902 | 11.785 |
| <b>Al</b>           | 4.158     | 4.118  | 4.3 52 | 4.102  | 4.012  | 4.01   |
| <b>Fe</b>           | 0.0076    | 0      | 0      | 0.011  | 0.0062 | 0.0179 |
| <b>Ti</b>           | 0.0088    | 0      |        | 0.012  | 0      | 0      |
| <b>Mg</b>           | 0         | 0      | 0      | 0.018  | 0.015  | 0.015  |
| <b>Na</b>           | 1.106     | 0.560  | 3.396  | 1.197  | 0.929  | 1.321  |
| <b>Ca</b>           | 0.121     | 0      | 0.095  | 0.110  | 0.104  | 0.165  |
| <b>K</b>            | 2.942     | 3.338  | 0.284  | 2.994  | 3.06   | 3.069  |
| <b>BaO</b>          | 0         | 0      | 0      | 0.011  | 0.038  | 0.005  |
| <b>Mol. percent</b> | <b>Or</b> | 70.57  | 85.63  | 7.52   | 69.61  | 74.76  |
|                     | <b>Ab</b> | 26.53  | 14.37  | 89.96  | 27.83  | 22.70  |
|                     | <b>An</b> | 2.90   | 0      | 2.52   | 2.56   | 3.62   |

### 3.6. Oxides

Oxide minerals of magmatic rocks consist of ilmenite, magnetite, hematite, corundum, spinel and ulvospinel. In order to do geothermometry the results indicate that they contain some geikielite ( $MgTiO_3$ ) and with a lesser amount pyrophanite ( $MnTiO_3$ ), Table No. 7.

**Table 7** EPMA results of oxides from granitoid rocks. Cations calculation is based on 6 Oxygens for ilmenite and 32 Oxygens for magnetite.

| Iron oxide   | Mag.          | Mag.             | Mag.             | Mag.             | Ilm.            | Ilm.             | I1m.             | I1m.             |
|--|---------------|------------------|------------------|------------------|-----------------|------------------|------------------|------------------|
| No   | Gkl           | Gk <sub>18</sub> | AD <sub>14</sub> | AD <sub>22</sub> | Gk <sub>1</sub> | Gk <sub>18</sub> | AD <sub>14</sub> | AD <sub>22</sub> |
| Oxides   |               |                  |                  |                  |                 |                  |                  |                  |
| TiO <sub>2</sub>   | 2.8           | 10.74            | 2.41             | 0.34             | 50.15           | 49.29            | 47.9             | 49.12            |
| SiO <sub>2</sub>   | 0.03          | 0                | 0.07             | 0.04             | 0.02            | 0.05             | 0                | 0                |
| FeO <sub>t</sub><br>{FeO<br>Fe <sub>2</sub> O <sub>3</sub> } | 28.3<br>68.12 | 77.86<br>6.21    | 26.73<br>69.09   | 30.69<br>68.28   | 38.8<br>3.67    | 37.78<br>4.98    | 38.96<br>2.31    | 39.25<br>2.08    |
| Cr <sub>2</sub> O <sub>3</sub>                               | 0.09          | 0.43             | 0.19             | 0.07             | 0.05            | 0.05             | 0.07             | 0.05             |
| Al <sub>2</sub> O <sub>3</sub>                               | 0.1           | 2.02             | 0.09             | 0.07             | 0               | 0.31             | 0.19             | 0.1              |
| V <sub>2</sub> O <sub>3</sub>                                | 0.38          | 0                | 0.28             | 0.29             | 0.16            | 0.24             | 0.12             | 0.09             |
| MnO  | 0.25          | 0.22             | 0.91             | 0.26             | 7.22            | 6.21             | 7.14             | 7.09             |
| MgO  | 0.06          | 2.14             | 0.04             | 0.04             | 0.12            | 1.45             | 3.23             | 2.14             |
| CaO  | 0             | 0.12             | 0                | 0                | 0.05            | 0.09             | 0.08             | 0.05             |
| Total  | 100.13        | 99.74            | 99.81            | 100. 8           | 100.32          | 100.42           | 100              | 99.97            |

| cations                                  |                 |                 |               |               |                |                |                |                |
|--|-----------------|-----------------|---------------|---------------|----------------|----------------|----------------|----------------|
| Ti                                       | 0.626           | 2.899           | 0.546         | 0.077         | 1.905          | 1.856          | 1.821          | 1.842          |
| Si                                       | 0.0087          | 0               | 0.02          | 0.012         | 0.001          | 0.0025         | 0              | 0              |
| Fe {Fe <sup>2+</sup><br>Fe <sup>3+</sup> | 7.034<br>15.251 | 23.453<br>0.261 | 6.77<br>15.74 | 7.88<br>15.77 | 1.643<br>0.139 | 1.582<br>0.186 | 1.647<br>0.085 | 1.638<br>0.078 |
| Cr                                       | 0.0214          | 0.129           | 0.043         | 0.016         | 0.002          | 0.0019         | 0.0028         | 0              |
| Al                                       | 0.35            | 0.865           | 0.032         | 0.025         | 0              | 0.018          | 0.01           | 0.059          |
| V  | 0.089           | 0               | 0.065         | 0.07          | 0.0064         | 0.0096         | 0.0048         | 0.0036         |
| Mg                                       | 0.026           | 1.146           | 0.018         | 0.018         | 0.0088         | 0.106          | 0.243          | 0.159          |
| Mn                                       | 0.0626          | 0.064           | 0.218         | 0.066         | 0.306          | 0.263          | 0.305          | 0.297          |
| Ca                                       | 0               | 0.043           | 0             | 0             | 0.0027         | 0.0048         | 0.0043         | 0.0026         |

#### 4. Geothermobarometry

The propose of geothermobarometry is to determine pressure and temperature (P&T) conditions of formation of rock [8]. During the past twenty years, laboratory experiments, thermodynamical models, calcultions, and analytical works by EPMA have provided a better situation to understand the P-T conditions of formation of minerals. One of the methods to indicate pressure of mineral crystallization in plutonic rocks is hornblende geobarometry [9]. According to this method, using this formula,  $p (\pm 3 \text{ kbar}) = -3.92 + 5.03 \text{ Al}^T$ , hornblendes in calc-alkalic rocks were crystallized under pressure of 2.28-6.33 kbar in the khezr-Abad pluton. The above calculated results are compared with results obtained by method of Johnson& Rutherford [10] , using the formula,  $P (\pm 0.5 \text{ kbar}) = -3.46 + 4.23 \text{ Al}^T$ , where  $\text{Al}^T$  is the total aluminum content reported as cations per 23 Oxygens formula unit. Typical hornblendes of dioritic rocks and other intermediate calc-alkaline rocks have a relative amount of  $X = \text{Mg}/(\text{Mg}+\text{Fe})$  close to 0.5, such a hornblendes have 1.5 Al atom in formula unit based on 23 Oxygens [11].

The above mentioned factor in hornblende bearing rocks of the studied area varies between 0.56-0.57 indicating and confirming calc-alkaline characteristics of them. To indicate pressure and temperature conditions of minerals and magmatic crystallization can be used Ab and An fractions of plagioclases and Ab and Or fractions in alkali-feldspars. In this order the following formulas and method were used [12, 13]:

$$KD = X^{Pl}_{ab} (ab+an) / X^{kf}_{ab} (ab+or)$$

$$Lnp = 11.2 \ln KD - 12.3$$

$$T^{\circ C} = 2080 / (\ln KD - 0.091 \ln p + 1.16)$$

$$P = \exp \left\{ \left[ \frac{\ln KD - (2080/T + 1.16)}{0.091} \right] \right\}$$

According to the above method, the temperatures of 900–920 °C are obtained for Khezr-Abad intrusive rocks. The magmatic pressure conditions for Khezr-Abad intrusive are 2.43 to 5.58 kbar. In addition to the above method, the method of Haselton et.al. [12], also is used.

$$T^{\circ C} = \frac{\left[ (X^{kf}_{or})^2 (18810 + 17030 X^{kf}_{ab} + 0.346 p) - (X^{Pl}_{an})^2 (28230 - 39520 X^{Pl}_{ab}) \right]}{10.3 (X^{kf}_{or})^2 + 8.314 \ln [(X^{Pl}_{ab})^2 (2 - X^{Pl}_{ab} / X^{kf}_{ab})]}$$

With this respect the temperatures of 810 to 867 °C are calculated for crystallization of feldspars of intrusive rocks. In addition to the thermometers mentioned above plagioclase-liquid geothermometry method also is used to identify magmatic condition temperatures.

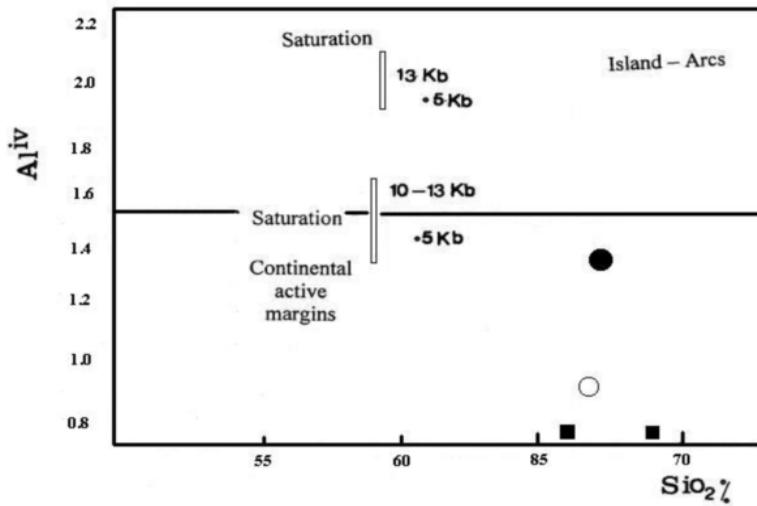
#### 4.1. Plagioclase-liquid geothermometry

This method is based on the equilibrium between the two phases. These two phases are rim of plagioclase crystal and the remaining liquid surrounding it. The calculated temperatures using this method are in the order of 805 to 837 °C.

#### 5. Physical conditions of crystallization in granitic magmas

By study of chemical composition of clinopyroxenes and amphiboles, it is possible to find out the amount of water vapour pressure and approximate percentage of water in magmas [14, 15, 16, 17 and 18].

On the basis of tetrahedral Al amounts in amphiboles against silica content of rock (Fig. 6), in addition to conditions of variations of water vapour pressure, magmatic amphiboles of Island-Arcs and continental active margins are recognizable from each other. According to (Fig. 6) magmatic amphiboles of the studied area are originated from magmas of continental active margins.



**Fig. 6** Limit of  $[Al^{IV}] = 1.5$  allows which amphiboles and therefore magmas of Island arcs are being recognized from amphiboles and magmas of continental active margins (the case in the study area) [18]. Open bars showing range of variations of synthetic amphiboles under pressures of 5 kbar [14].

○: Quartz-diorite. ■: Granite. ●: Granodiorite.

## 6. Discussion and conclusions

Plutonic rocks of Khezr-Abad are intruded into the Taft limestones. Strong evidence to confirm this matter comes from the occurrence of skarn formation in the limestones. Petrographic evidence show that metamorphic phenomena including recrystallization of limestones and tuffs, results in formation of marbles and skarn. Electron microprobe studies of minerals show that Fe-Mg minerals (pyroxene, amphibole, micas) are Mg rich.

Temperature and pressure identification of plutonic rocks emplacement are based on different methods. The obtained temperatures and pressures are from 810 – 867 °C and 2.43 to 5.58 kbars respectively.

General conclusions are as the following:

- 1- With respect to petrography, the studied granitoids show a wide varieties, ranging from diorite to alkali-granite, indicating differentiation process in generation of the rocks .
- 2- The host-rocks were cretaceous limestones, and have changed to skarn. Main minerals of the skarn are diopside, garnet (andradite), phlogopite, scapolite, talc and serpentine.

- 3- Known textures in magmatic rocks are granular, granophyric (graphic), perthite, anti-perthite and sieve textures. The graphic textures indicate eutectic or cotectic crystallization of quartz and alkali-feldspars .
- 4- The essential minerals of granitoids consist of quartz, plagioclase (albite, oligoclase and andesine), orthoclase, mica (biotite and with a lesser amount muscovite). Plagioclases show zoning. This phenomenon indicates rapid depressurizing and thermodynamical and chemical variations of crystallization environment .
- 5- Textural (perthite and granophyric), mineralogical (zoned texture of plagioclases), chemical (range of  $\text{SiO}_2$  variations), low K/Na ratio, high Ca content, lack of aluminum minerals and muscovite evidences, and the presence of intermediate thermals such as diorite and granodiorite all suggest a nature for the granitoids .
- 6- The results of EPMA analyses of pyroxenes show that they are high calcic, and of augite and diopside types.
- 7- On the basis of  $\text{Mg}/(\text{Mg} + \text{Fe}^{2+})$  and Si, amphiboles of magmatic rocks are magnesiohornblende and actinolite.
- 8- According to EPMA studies, micas are biotite in plutonic rocks and phlogopite in metamorphic rocks .
- 9- Plagioclases in plutonic rocks are mostly oligoclase - andesine , and alkali feldspars are orthoclase, microcline or perthitic orthoclase or microcline and albite.
- 10- With regard to  $X_{\text{ab}}^{\text{pl}}$  (ab + an) ratios, it is suggested that rocks were crystallized at moderate to low depths of the crust .
- 11- Geochemical analysis and geothermobarometry studies indicate a pressure and temperature of 2.43-5.5 kbars and 810–867 °C respectively.
- 12- With respect to broad presence of perthite and anti-perthite textures and plagioclase-liquid geothermobarometry studies, the water vapour pressure in the intrusive masses are probably 1 to 2 kbar .
- 13- Plot of  $\text{Al}^{\text{IV}}$  against  $\text{SiO}_2$  on the diagram shows that amphiboles of magmatic rocks of the studied area related to magmas of continental active margins .

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